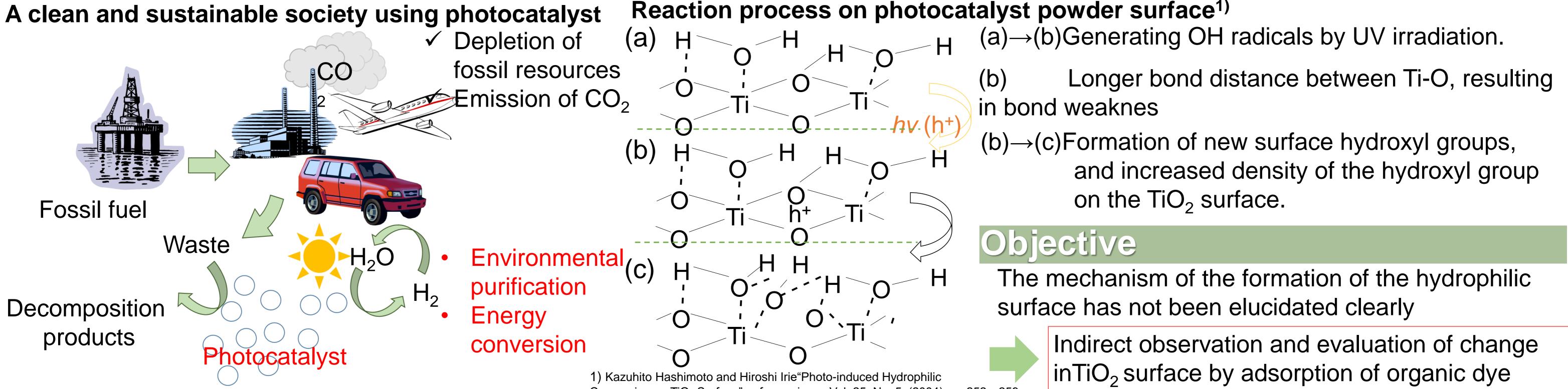
Hynes, LEVEL1, Hall B MRS Fall Meeting ET07.09.06 November 28, 2018 Spectroscopic Study for Hydrophilic Surface of TiO₂ Photocatalysts Modified with Au Nanoparticles

Hayato Kondo, Yosuke Kageshima, Hiromasa Nishikiori

¹Department of Engineering, Graduate School of Science and Technology, Shinshu University, Nagano, Nagano, Japan

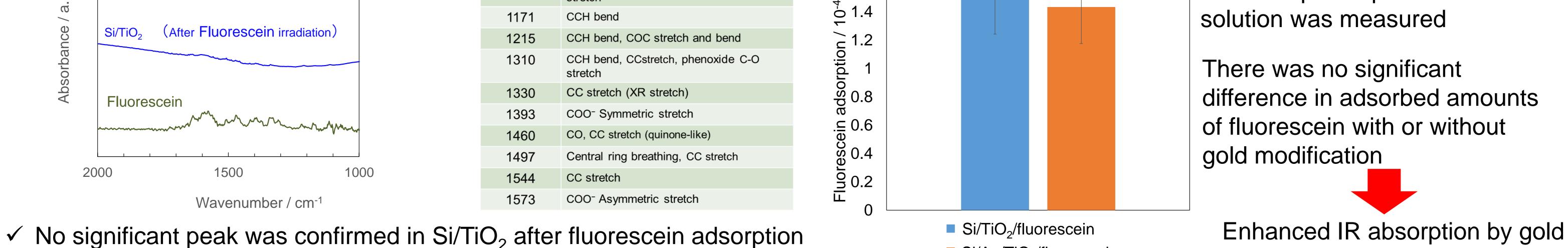
Introduction



Conversion on TiO₂ Surface" surface science Vol. 25, No. 5, (2004) pp. 252–259.

Evnerimental

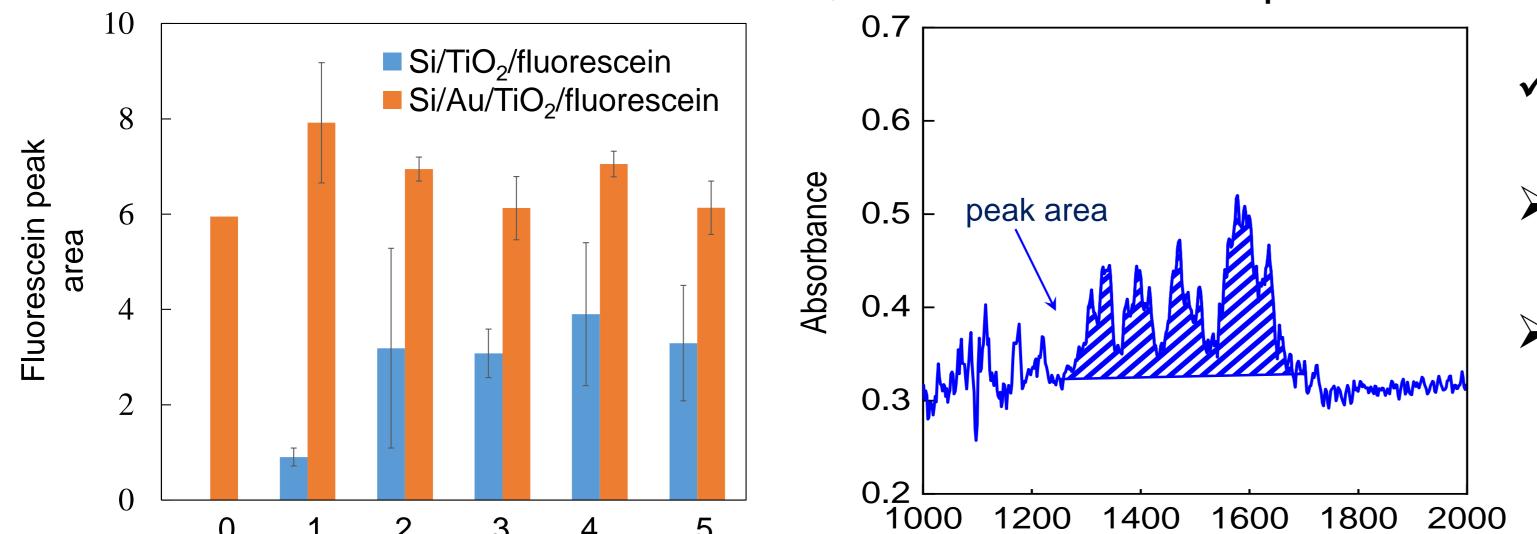
Experimental				
Si substrate	Experimental setup for measurements of light irradiation time dependence			
SISUDSIIALE	Deposition of Au nanoparticle/light source (15 W BI		rce (15 W Black light)	
	by vacuum evaporation			
Si/Au	Sol-gel method		adiation time : 60~300 min	
	Titania sol	Distan	Distance from light source : 10 cm	
	Ethanol, Nitric acid	10 cm	Fluorescein	
	Distilled water, TTIP	1. Light irradiation	HO \sim 0 \sim 0 \sim 0 \sim 0	
Si/Au/TiO ₂	773 K, 30 min heat treatment	2. Immersion in dye solution Si or Si/Au 3. Drying in air Ade	sorbed	
	• 60~300 min Light			
	irradiation		$\begin{array}{c} & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & & \\ & & \\ & & & \\ & & \\ & &$	
Si/Au/TiO ₂ /Fluorescein	 Fluorescein 1.0×10⁻² M 	TiO_2 thin film	pKa = Monoanion pH5.4 6.4 Dianion pH9.1	
	60 min Immersion			
Result and Discussio	Ŋ			
Observation of adsorbed dye on TiO ₂ surface by FT-IR Adsorbed amounts of fluorescein on TiO ₂				
Peak position of each vibration mode ¹⁾ 2				
Si/Au/TiO ₂ (After Fluorescein irraiation)	Wavenum Vibration mode ber / cm ⁻¹	ຮັ້ 1.8	redissolved in 1.0×10^{-2} M HCl and	
	1115 Asymmetric CCH bend, C-O stresstretch	etch, C-C 2 1.6	the absorption spectrum of the	
/a.i	1171 CCH bend	⁴ 01.4	colution was masured	



 \checkmark A peak of adsorbed fluorescein was confirmed at 1300-1700 cm⁻¹ in Si/Au/TiO₂

Changes in amounts of adsorbed dye according to light irradiation time

Time courses of adsorbed amounts of dye Definition of the peak area



 \checkmark In the case of Si/TiO₂, the peak area of the IR absorption peak of fluorescein is increased by UV irradiation time longer

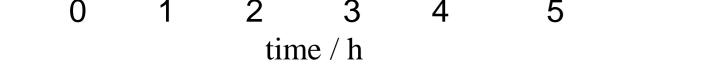
Si/Au/TiO₂/fluorescein

- \checkmark In the case of Si/Au/TiO₂/fluorescein, regardless of UV irradiation time, the peak area of the dye is almost constant
- Changes in actual surface conditions may be completed within 1 hour of light irradiation
- Enhanced surface plasmon enhancement of gold allowed accurate tracking of adsorbed dye

Conclusion

Significant IR absorption of fluorescein

surface plasmon resonance



No significant change in spectral

shape could be confirmed at 1

and 5 hours of light irradiation

considered to be in the form of

Almost all adsorbed dyes are

Observation for adsorbed dye in a

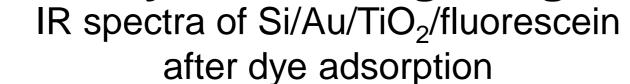
shorter time scale should be is

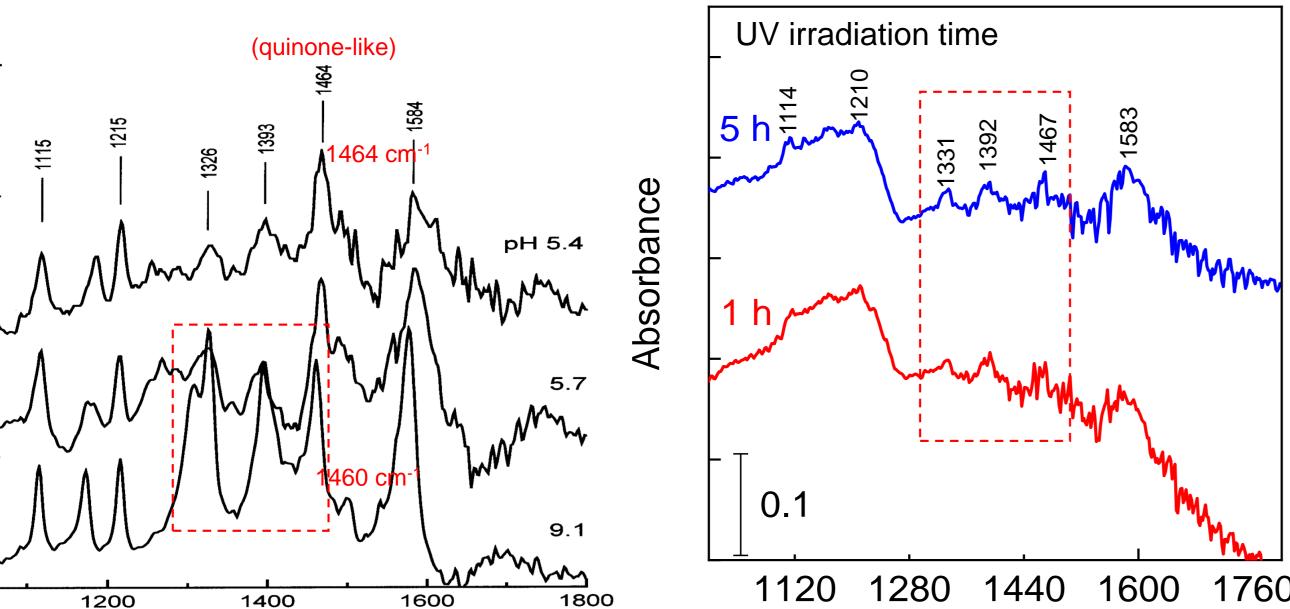
Wavenumber / cm⁻¹

Changes in IR spectra of adsorbed dye according to light irradiation time

IR spectra of fluorescein aqueous solution at different pH²)

Wavenumber (cm⁻¹)





2) L. Wang et al., Spectrochimica Acta Part A, 57 (2001) 1781-1791. Wavenumber / cm⁻¹

dianion

necessary

could be observed by-with using gold modified Si/TiO₂

- There was no significant difference in adsorption amount regardless of the gold modification
- Fluorescein could be observed by surface plasmon enhancement of gold
- Regardless of UV irradiation time, the peak area of adsorbed dye was almost constant
- From the shape of the IR spectrum, almost all adsorbed dyes are considered to be in the form of dianion
- → Investigation of adsorbed dye in a shorter time and detailed observation of temporal change in surface condition are future issues